Use of mass spectrometry in the study of enzymes

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Overview of class

- Modification of the enzyme to regulate its activity
- Examining the chemistry of enzyme:substrate intermediates
 - Locating the site of inactivation of suicide inhibitors
- Reaction mechanism
 - Measuring all substrates and products
 - Enzyme kinetics
 - Structural studies
 - Stopped flow
- BAT, my kinda 'zyme Erin Shonsey/Tyler Stewart

A good review of this topic

 Kelleher, NL and Hicks LM. Contemporary mass spectrometry for the direct detection of enzyme intermediates. Current Opinion in Chemical Biology 9: 424-430, 2005.

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Mass spectrometry and the study of enzymes

- Enzymes often undergo posttranslational modifications in order to be active under the conditions in a cell
 - for example, many enzymes in the signal transduction pathways are activated by phosphorylation on serine, threonine and tyrosine residues
 - EGF receptor (tyrosine kinase), TGF beta type I receptor (serine kinase)
 - sites of phosphorylation can be determined by mass spectrometry because of the increase in mass of m/z 80 of peptides containing each phosphate group

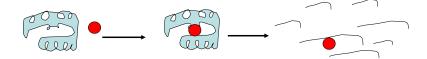
Enzymes and Mass Spec

Enzymes may undergo changes in structure once activated (see above) or during the reaction they catalyze

- this could be probed by H-D exchange experiments (this is coming up in Peter Prevelige's lectures)
- simulation of phosphorylation may be necessary by mutating serine and threonine groups to aspartate and glutamate, respectively

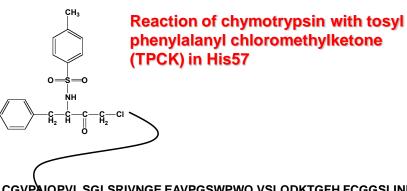
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MS of enzymes



Enzymes can be inactivated by suicide substrates - these come into the active site and undergo a covalent reaction, thereby blocking the approach of other substrate molecules

 to locate the region of the enzyme to which the suicide substrate is bound, carry out a trypsin digest and look for a peak that has undergone a molecular weight change (consistent with the structure of the suicide inhibitor)



CGVPAIQPVL SGLSRIVNGE EAVPGSWPWQ VSLQDKTGFH FCGGSLINEN 50

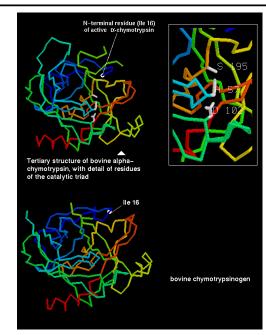
WVVTAAHCGV TTSDVVVAGE FDQGSSSEKI QKLKIAKVFK NSKYNSLTIN 100

NDITLLKLST AASFSQTVSA VCLPSASDDF AAGTTCVTTG WGLTRYTNAN 150

TPDRLQQASL PLLSNTNCKK YWGTKIKDAM ICAGASGVSS CMGDSGGPLV 200

CKKNGAWTLV GIVSWGSSTC STSTPGVYAR VTALVNWVQQ TLAAN

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Chymotrypsin has a catalytic triad consisting of Asp102, His57 and Ser195

His57 is the site of reaction of chymotrypsin with TPCK to form a stable covalent bond, thereby acting as a suicide inhibitor

Possible proteases for locating TPCK-peptide

CGVPAIQPVL SGLSRIVNGE EAVPGSWPWQ VSLQDKTGFH FCGGSLINEN

WVVTAAHCGV TTSDVVVAGE FDQGSSSEKI QKLKIAKVFK NSKYNSLTIN

NDITLLKLST AASFSQTVSA VCLPSASDDF AAGTTCVTTG WGLTRYTNAN 150

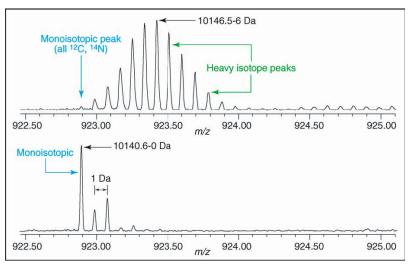
TEOPLANDASI-BLISHTNICKE EXVECTSWERMY VSAGDRAGER FCGGSLINEN

<mark>čikkngamely gyspygsage</mark> bdatessyerr ytalybkvfragikaabliin 100

NDITLLKLST AASFSQTVSA VCLPSASDDF AAGTTCVTTG WGLTRYTNAN

SMGPSGGFILY 600TTSDVVVAGE FDQGSSSEKI QKLKIAKVFK AAGTTCVTTG WGLTRYTNAN 150 TPDRLQQASL PLLSNTNCKK YWGTKIKDAM ICAGASGVSS CMGDSGGPLV 200 CKKNGAWTLV GIVSWGSSTC STSTPGVYAR VTALVNWVQQ TLAAN Stephen Barnes BMG 744 02-13-12

Advantage of growing recombinant bacteria on ¹²C/¹⁴N labeled substrates - protein is close to monoisotopic mass



Kelleher & Hicks, 2005 Stephen Barnes BMG 744 02-13-12

Mass spectrometry and enzymecatalyzed reactions

In the simplest case, an enzyme (E) reacts with a substrate (S) - an intermediate complex is formed (ES) and it is converted to an enzyme: product complex (E:P) before the product dissociates.

$$E + S \rightarrow ES \rightarrow EP \rightarrow E + P$$

First order reaction - some second order reactions behave like a first order reaction when there is an excess of one substrate and the conversion of the other is <10%.

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Mass spectrometry and enzymecatalyzed reactions

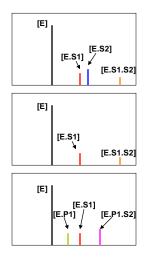
E.S1 E.S1.S2
$$\rightarrow$$
 E.P.Q E.S.2 E.Q

By measuring the molecular weights of the forms of the enzyme:substrate (product) complexes, mass spectrometry can throw enormous light on the mechanism

Mass spectrometry and enzymecatalyzed reactions

More typical reactions involve two substrates (S1 and S2) and two products (P1 and P2). The problem in this case is the order of addition

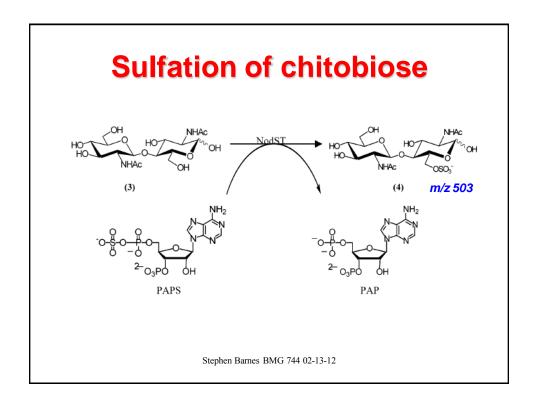
- is it a random mechanism? If so, both E.S1 and E.S2 exist
- is it an ordered mechanism? In this case, S1 has to bind first. So, there will be E.S1 and E.S1.S2, but no E.S2
- is it a Ping-Pong mechanism? In this case, E.S1→E.P1 before S2 binds to form E.P1.S2



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Mass spectrometry and substrates and products of enzyme reactions

- Most enzyme reactions are studied by measuring the appearance of a product or (more rarely) the disappearance of a substrate
- If the substrate or product has a unique absorbance or fluorescence, the reaction can be followed in real time
- Some substrates have no usable absorbance or fluorescence - these can be measured using a radiolabeled substrate - the product is isolated by a solvent extraction procedure, or by HPLC or TLC. These reactions cannot be observed in real time
- Mass spectrometry has the advantage that it is capable of measuring all substrates and products, as well as the enzyme itself



Set up for the ST assay

Pi et al., Biochemistry 41:13283

- NodST purified by Ni-affinity chromatography
 - dialyzed against 100 mM Tris-HCl, pH 8.0 20 mM β-ME
 - Diluted into 10 mM NH₄Ac buffer, pH 8.0
- Incubate (25 μ I) quenched with 100 μ I of MeOH containing internal standard
- Diluted incubate (40 μl) introduced into ESI source at 20 μl/min
- MS on a ThermoFinnigan LCQ monitoring m/z 503 and m/z 468 (internal standard)

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Kinetics of chitobiose ST by ESI-MS

Pi et al., Biochemistry 41:13283

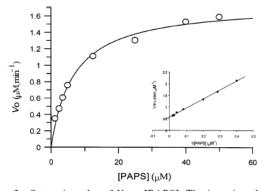


FIGURE 3: Saturation plot of V_0 vs [PAPS]. The inset is a double-reciprocal plot of $1/V_0$ vs 1/[PAPS] ([PAPS] = 1.25, 2.5, 3.5, 5, 12.5, 25, 40, and 50 μ M, [chistobiose] = 1 mM, [NodST] = 90 nM, and pH 8.0).

Inhibition of ST by PAP using ESI-MS

Pi et al., Biochemistry 41:13283

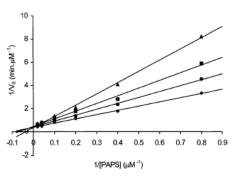
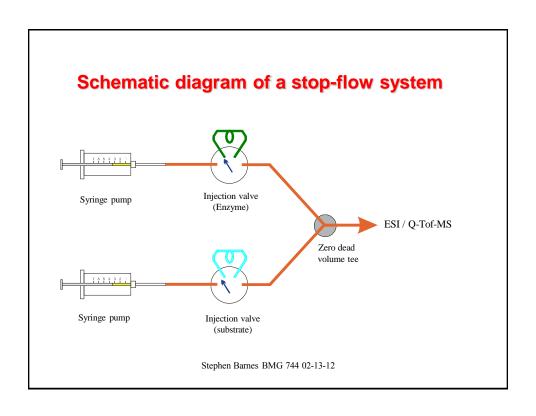


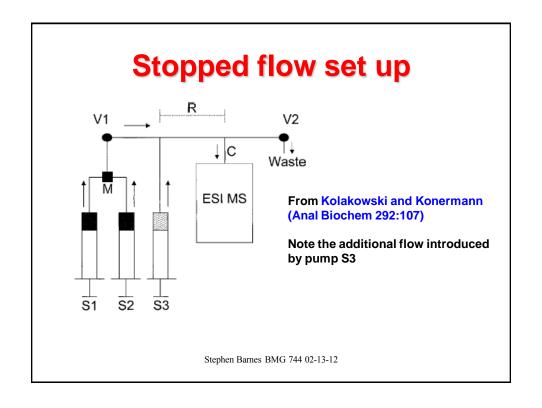
FIGURE 5: Double-reciprocal plot of $1/V_0$ vs 1/[PAPS] at different PAP concentrations: $0 \mu M$ PAP (\blacktriangle), $0.75 \mu M$ PAP (\blacksquare), $1.5 \mu M$ PAP (\blacksquare), and $3.0 \mu M$ PAP (\spadesuit) ($[PAPS] = 1.25, 2.5, 5, 10, 25, and <math>50 \mu M$, [chitobiose] = 1 mM, [NodST] = 90 nM, and pH 8.0).

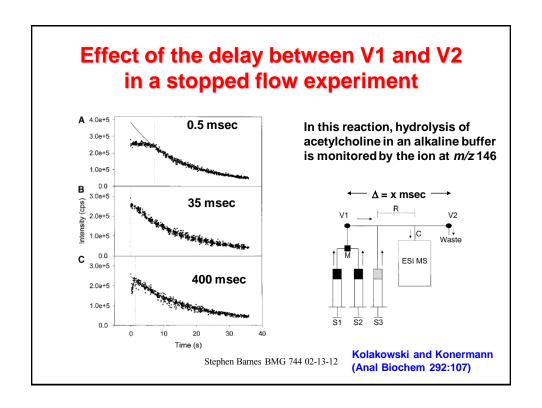
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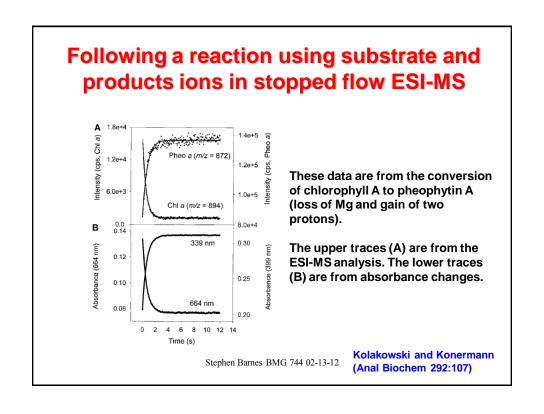
Non-covalent enzyme:substrate complexes

- Shifting the enzyme from neutral pH conditions to the acidity of the spraying solution may break down the complex
- Spraying at neutral pH will increase the observed m/z values (the protein is less charged with protons)
- The larger m/z ions can be observed with an electrospray-TOF or a Qq TOF

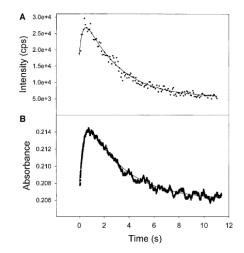








Unfolding kinetics of myoglobin by stopped-flow ESI-MS



The upper trace (A) is the 14th charge state of holo-myoglobin [M+14]¹⁴⁺ (m/z = 1255.9)

The reaction is created by a pH jump from 6.0 to 3.0. The lower trace (B) is the absorbance at 441 nm.

The estimated time constants for the bi-exponential process are 0.29/2.8 sec for A and 0.33/3.1 sec for B

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Kolakowski and Konermann (Anal Biochem 292:107)

Summary of the use of (real time) ESI-MS to follow enzyme reactions

The pros:

- All the substrates and products (as well as the enzyme itself) can be studied simultaneously
- It's applicable to compounds with no absorbance or fluorescence

The cons:

- The buffer for the reaction has to be chosen very carefully
- Ammonium salts are the best candidates, but they may have an effect on the reaction rates

Studying multiple enzymes simultaneously

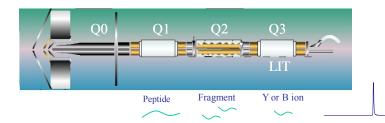
- Old style of research
 - Measure one purified enzyme at a time
- New style of research
 - Measure all the enzymes at the same time
 - Equivalent to study of a signaling pathway (many are phosphorylation steps)

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Multiple reaction ion monitoring

- This technique allows us to measure the proteins as well as the substrate and products
 - Based on selection of the parent ion, fragmentation, and selection of specific fragment ions

Multiple Reaction Monitoring



- High specificity and sensitivity for detecting components in a complex mixture
- Largest linear dynamic range for quantitation
- Well accepted as the MS technique for quantification (small molecule world)
- Triple quadrupole or Q TRAP® system mass spectrometers

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Drug analysis in industry

- The reality is that drug targeting and drug development eventually encounter the realities of metabolism
- A major activity in the post discovery phase is to determine the metabolic reactivity of the drug, particularly the cytochrome P450 system

Multiplexed analysis of the drug metabolizing enzymes

Approach	Advantages	Disadvantages
Cocktail strategy	Several activities in a single experiment Reduced number of samples	
	Less time and expense	Probe-probe interaction
	Enhanced throughput Rapid microsomal characterization	Ion suppression Metabolism overlapping
	Rapid phenotype of tissues	Wetabolisiii overlappilig

Individual

strategy

Selective CYP activity

Avoids overlapping metabolism

Avoids probe-probe interaction

Lahoz et al., 2008

Large number of

Time consuming

strategies

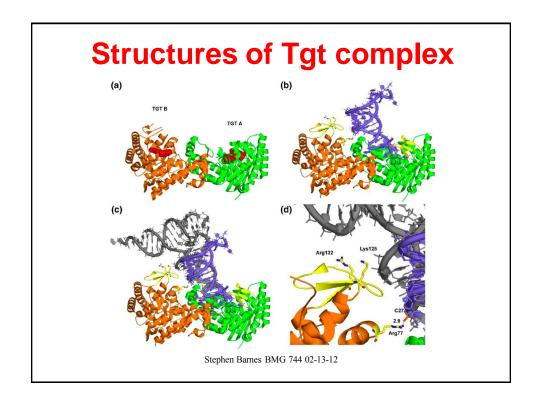
analysis

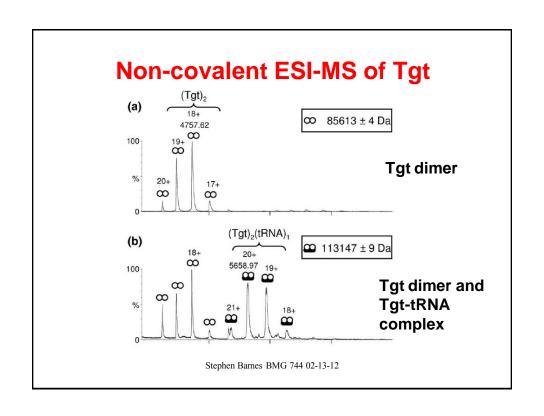
tRNA-guanine transglycosylase

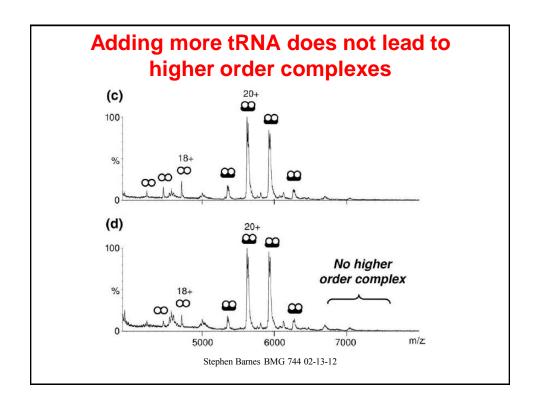
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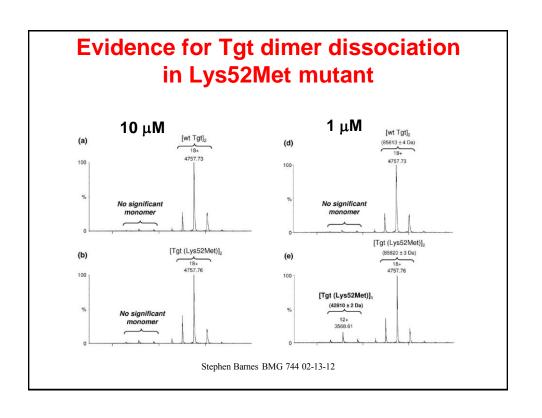
- This enzyme (Tgt) modifies tRNAs and may be a target for antibiotics
- · The enzyme forms a dimer with tRNA
 - Can we observe the dimer using mass spectrometry?
 - Can we explore the dimer interface by carrying out mutations of Tgt part of the interface?

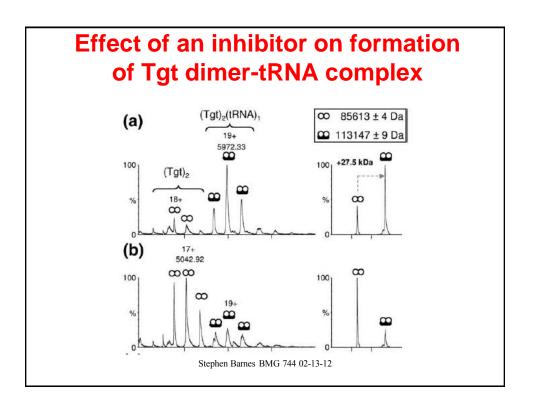
Ritschel et al. J Mol Biol (393:833-847, 2009)





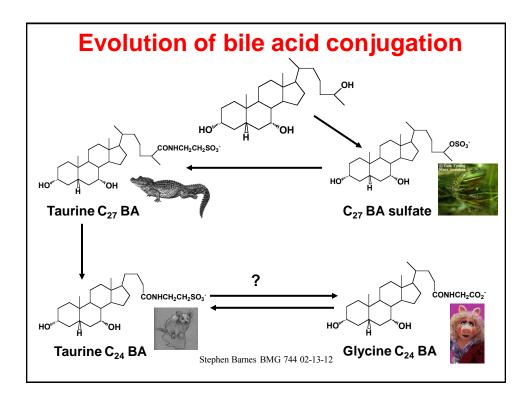




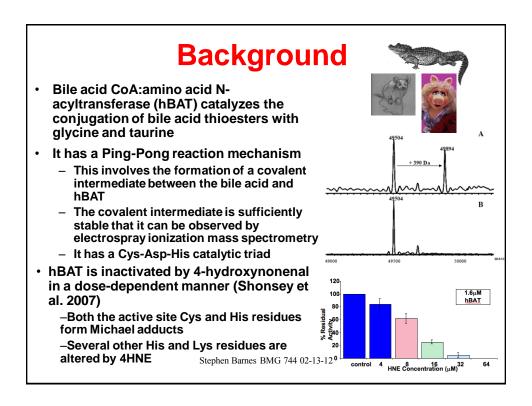


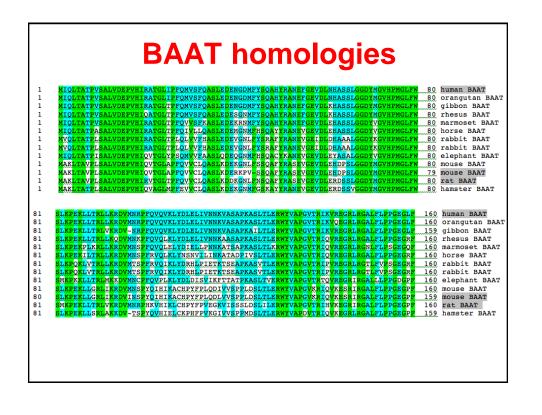
Conclusions about non-covalent complexes by ESI-MS

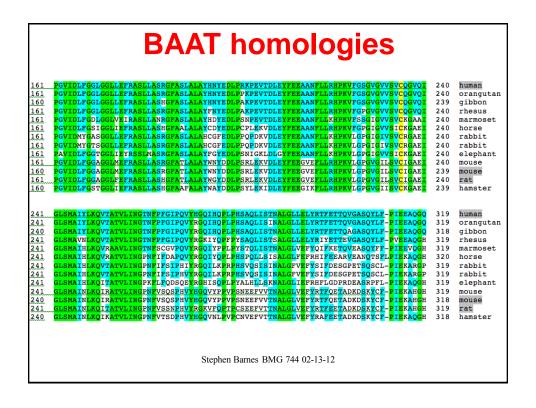
- The physical state of the enzyme with its substrates and inihibitors can be readily observed and quantified by ESI-MS
- Requires sufficient protein (in the Ritschel study they used 1-10 μM – for a 25 kDa protein, this would be 25-250 mg/L or 25-250 μg/ml)
- The spraying medium is important in this case 500 mM ammonium acetate, pH 8

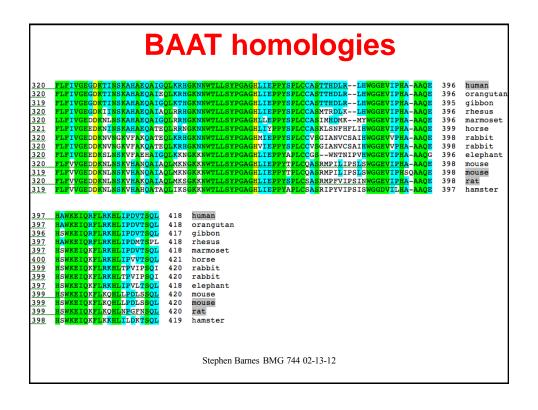


Bile acid N-acylamidate formation (in hepatocytes)



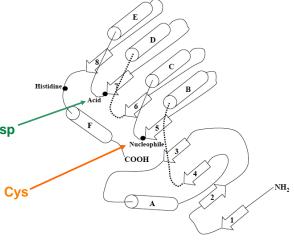






hBAT is a member of a class of proteins with an α/β hydrolase fold

The regions of hBAT that were part of α -helices, β -sheets and unstructured regions were determined by the use of the JUFO (www.Meilerlab.org)



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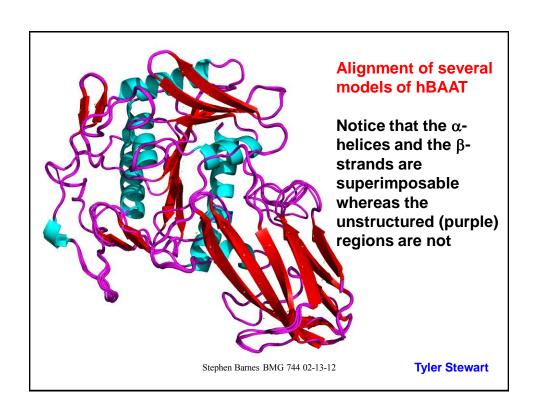
Modeled structure of hBAT

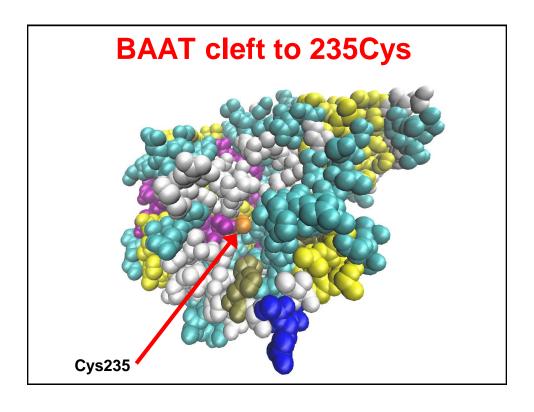
In the absence of hBAT crystals, structural homology experiments using FUGUE were performed. These identified two previously crystallized proteins, ACOT2 and ACOT4 as structural homologs of hBAT. These two structures were used to create a threaded structure for hBAT which was refined using Modeller 9v10

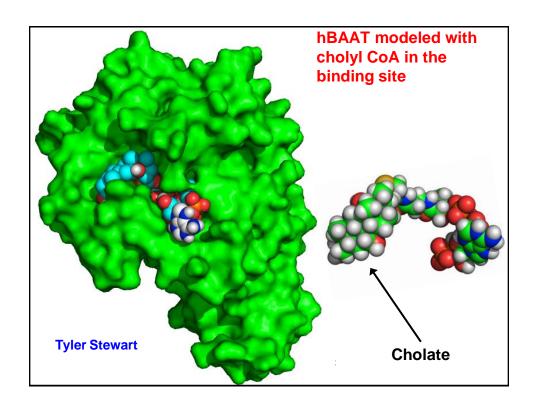
His362
Asp328

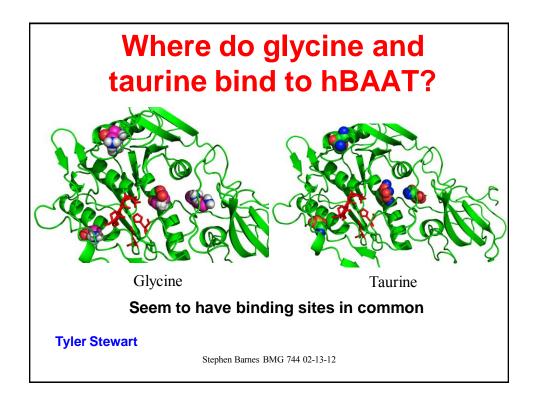
Tyler Stewart

http://tardis.nibio.go.jp/fugue/prfsearch.html

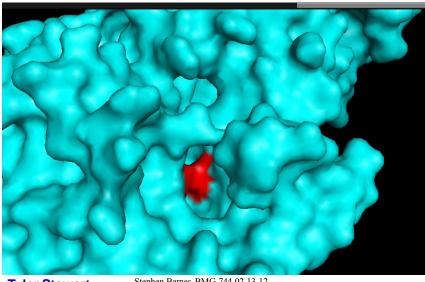










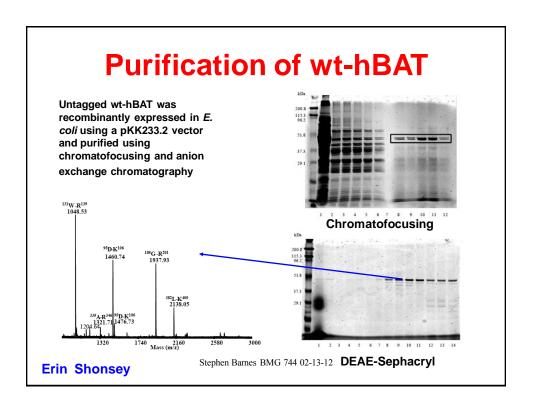


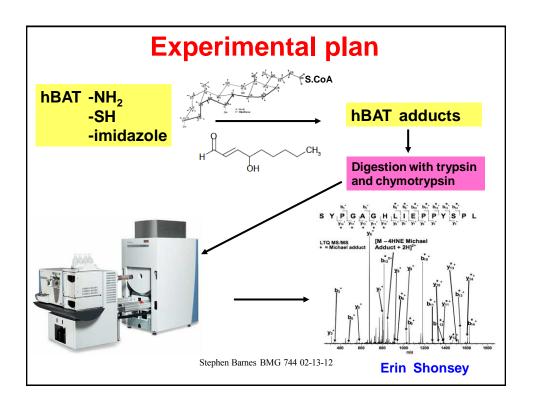
Tyler Stewart

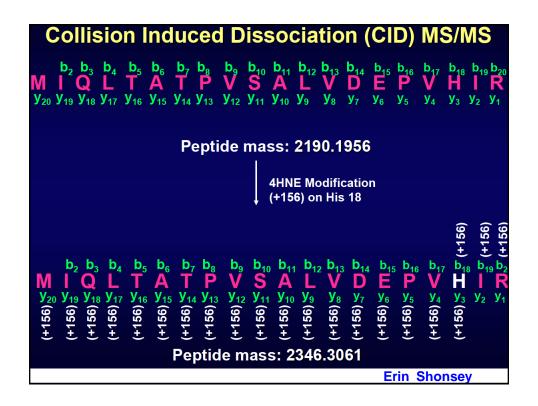
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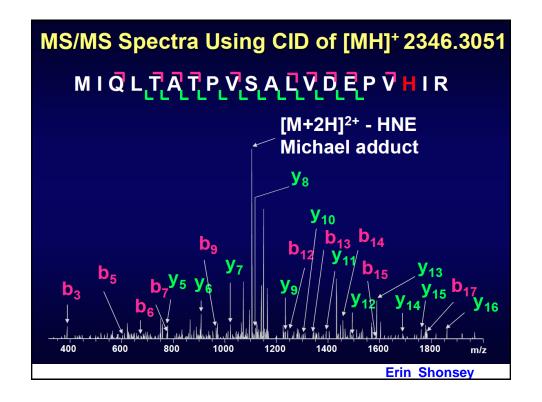
Hypothesis

- The ability of 4HNE to react with Cys, His and Lys residues is dependent on their accessibility to solvent containing 4HNE.
- · Therefore, in the presence of cholyl CoA, and hence covalently linked cholate to Cys235, the accessibility of 4HNE to the active site will be blocked, thereby revealing the residues that are involved









Concentration-dependent sites of 4HNE modification on hBAT

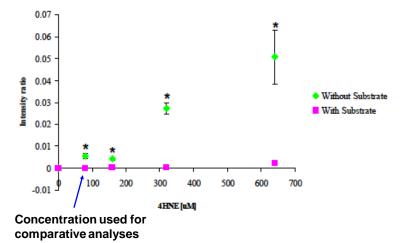
RЛ	odif	iad	۸n	ain.	٠,	٠i	a
IVI	oan	1eca	AII	11116) A	ш	a

Peptide	128 µM HNE	64 μM HNE	32 μM HNE	16 µM HNE	8 µM HNE
A <mark>H</mark> AEQAIGQLKR	H336	H336	H336	H336	H336
RLHWGGEVIPHAAAQEHAWK	H397	H397	H397	H397	
AQGQFLFIVGEGD <mark>K</mark> TINS K	K329, K334	K329, K334	K329, K334	K329, K334	K329, K334
MIQLTATPVSALVDEPVHIR	H18	H18			
RANEFGEVDLNHASSLGGDYMGV HPMGLFWSLKPEK	H62, H74	H62, H74	H62, H74	H62, H74	H62
HGQIHQPLPHSAQL	H271, H274, H279	H271, H274, H279	H271, H274, H279	H271, H279	H271, H279
NNWTLLSYPGAGHLIEPPYSPLCCA STTHDLR	H362, C372, C373, H378	H362	H362	H362	H362

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Shonsey et al., 2007

Cholyl CoA blocks 4HNE adducts on the His383 residue of hBAT



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Erin Shonsey

